### 2-(2,4-Cyclopentadien-1-ylidene)adamantane\*

By J. GABRIEL GARCIA, MARK L. MCLAUGHLIN<sup>†</sup> AND FRANK R. FRONCZEK

Department of Chemistry, Louisiana State University, Baton Rouge, Louisiana 70803-1804, USA

(Received 8 September 1988; accepted 3 January 1989)

CI

C2 C3 C4

C5 C6 C7 C8

C9 C10

C11

C12

C13 C14

C15

Abstract.  $C_{15}H_{18}$ ,  $M_r = 198 \cdot 3$ , monoclinic,  $P2_1/n$ , a = 6.942 (2), b = 15.824 (4), c = 10.251 (2) Å,  $\beta =$  $V = 1126 \cdot 0$  (8) Å<sup>3</sup>, 90⋅89 (2)°, Z = 4, $D_{\rm r} =$  $1 \cdot 170 \text{ g cm}^{-3}$ ,  $\lambda(\text{Mo } K\alpha) = 0 \cdot 71073 \text{ Å}$ ,  $\mu = 0.61 \text{ cm}^{-1}$ F(000) = 432, T = 296 K, R = 0.053 for 1388 observations (of 1977 unique data). The bond lengths of the 1,3-cyclopentadienylidene and adamantane rings are normal. The bond angle exo from the cvclopentadienylidene ring is 111.5 (2)°. This bond angle closely mimics the analogous bond angle of a difulvene that exhibits an enhanced absorbance [McLaughlin, Cronan & Fronczek (1988). Acta Cryst. C44, 1747-1749]. The cyclopentadienylidene ring is planar, with maximum deviation of 0.002 (3) Å.

Experimental. The title compound was prepared as described earlier (Olah, Surya & Liang, 1977), m.p. 361–363 K, crystal size  $0.30 \times 0.40 \times 0.45$  mm. Space group from diffraction symmetry 2/m and systematic absences hol with h+l odd and 0k0 with k odd. Cell dimensions from setting angles of 25 reflections having  $13 > \theta > 11^{\circ}$ . Data collected on Enraf-Nonius CAD-4 diffractometer, Mo Ka radiation, graphite monochromator,  $\omega$ -2 $\theta$  scans designed for  $I > 25\sigma(I)$ , subject to max. scan time = 120 s, scan rates varied  $0.91-4.0^{\circ}$  min<sup>-1</sup>. Data having  $1 \le \theta \le$ 25°,  $0 \le h \le 8$ ,  $0 \le k \le 18$ ,  $-12 \le l \le 12$  measured and corrected for background, Lorentz, polarization, and decay. Three standard reflections (110, 020, 002) decreased in intensity by 12.3% during data collection, apparently due to sublimation. A linear correction was applied. No absorption corrections were made. Redundant 0kl and  $0k\bar{l}$  data were averaged,  $R_{int} = 0.021$ . Structure solved by direct methods, refined by full-matrix least squares based upon F, using 1388 data for which  $I > 1\sigma(I)$ , weights  $w = 4F_o^2[\sigma^2(I) +$  $(0.02F_o^2)^2$ ]<sup>-1</sup> using Enraf-Nonius SDP (Frenz & Okaya, 1980), scattering factors of Cromer & Waber (1974), anomalous-dispersion coefficients of Cromer (1974), 589 unobserved data. Non-H atoms refined anisotropically; H atoms located by  $\Delta F$  syntheses and refined isotropically. Final R = 0.053 (0.093 for all 1977 data), wR = 0.046, S = 1.726 for 209 variables. Maximum shift  $0.01\sigma$  in the final cycle, max. residual density 0.16, min.  $-0.16 \text{ e} \text{ Å}^{-3}$ , extinction coefficient  $g = 8.1 (11) \times 10^{-7}$  where the factor  $(1 + gI_c)^{-1}$  was applied to  $F_c$ . The fractional coordinates of the title

# Table 1. Atomic coordinates and equivalent isotropic thermal parameters

8 2/77

---



Fig. 1. 2-(2,4-Cyclopentadien-1-ylidene)adamantane.



Fig. 2. ORTEP drawing (Johnson, 1965) of the molecule, representing heavy atoms as 40% probability ellipsoids and H atoms as circles of arbitrary radius.

© 1989 International Union of Crystallography

<sup>\*</sup> Adamantane is tricyclo[3.3.1.1<sup>3,7</sup>]decane.

<sup>&</sup>lt;sup>†</sup>Author to whom correspondence should be addressed.

Table 2. Bond distances (Å), angles (°), and selectedtorsion angles (°)

C1 C2 C6 C7 C9 C12	C2 C3 C7 C12 C15 C13	1-459 1-327 1-506 1-533 1-528 1-531	0 (2) 7 (3) 5 (2) 8 (3) 8 (3) 1 (3)	C1 C3 C6 C8 C10 C13	C5 C4 C11 C9 C11 C14	1.45 1.45 1.51 1.53 1.52 1.52	9 (2) 1 (3) 0 (2) 2 (3) 2 (3) 6 (3)		C1 C4 C7 C9 C11 C13	C6 C5 C8 C10 C14 C15	1.34 1.33 1.52 1.52 1.54 1.51	2 (2) 7 (3) 9 (2) 6 (3) 1 (3) 9 (3)	
C2 C1 C7 C8 C8 C6 C7 C14	C1 C2 C5 C6 C7 C9 C11 C12 C13	C5 C3 C4 C11 C12 C15 C10 C13 C15	104.8 (2) 109.1 (2) 108.3 (2) 111.5 (2) 109.4 (2) 109.2 (2) 109.2 (2) 109.6 (2) 109.8 (2)	C2 C2 C1 C6 C7 C10 C6 C12 C11	C1 C3 C6 C7 C8 C9 C11 C13 C14	C6 C4 C7 C8 C9 C15 C14 C14 C13	127.9 108.6 124.3 109.7 109.5 109.6 108.2 108.7 109.6	(2) (2) (1) (2) (2) (2) (2) (2) (2)	C5 C3 C1 C6 C8 C9 C10 C12 C9	C1 C4 C6 C7 C9 C10 C11 C13 C15	C6 C5 C11 C12 C10 C11 C14 C15 C13	127- 109- 124- 108- 108- 110- 109- 109- 109-	2 (2) 1 (2) 2 (2) 3 (2) 8 (2) 3 (2) 2 (2) 7 (2) 9 (2)
C5 C1 C11 C1 C1 C7	C1 C6 C6 C6 C6	C6 C7 C7 C11 C11	C7 C8 C8 C10 C10	-0 -123 58 123 -58	·3 (4) ·0 (2) ·6 (2) ·2 (2) ·4 (2)		C2 C1 C11 C1 C1 C7	C1 C6 C6 C6 C6	C6 C7 C7 C1 C1	C C C 1 C 1 C	11 12 12 - 14 14	0.5 117.7 -60.8 118.1 60.4	(4) (2) (2) (2) (2)

compound are given in Table 1. Fig. 1 is a line drawing and Fig. 2 is a perspective drawing showing the atom numbering. Bond distances, angles, and selected torsion angles are presented in Table 2.\*

Related literature. The structures of dimethylfulvene at 248 K (Norman & Post, 1961) and the difulvene, 1,4-bis(2,4-cyclopentadien-1-ylidene)cyclohexane (Mc-Laughlin, Cronan & Fronczek, 1988), exhibit the

expected single-double bond alternation within the fulvene ring system like the title compound. The bond angle C7-C6-C11 = 111.5 (2)° of the title compound compares with the analogous bond angles of dimethyl-fulvene, 114.0 (6)°, and 1,4-bis(2,4-cyclopentadien-1-vlidene)cyclohexane, 112.71 (9)°.

The adamantane ring system has been extensively characterized including the parent hydrocarbon by Nordman & Schmitkons (1965) and refined by Donohue & Goodman (1967). The geometry about the double bond of adamantylideneadamantane (Swen-Walstra & Visser, 1971) is similar to that of the title compound.

We thank the donors of the Petroleum Research Fund, administered by the American Chemical Society, for support of this research.

#### References

- CROMER, D. T. (1974). International Tables for X-ray Crystallography, Vol. IV, Table 2.3.1. Birmingham: Kynoch Press. (Present distributor Kluwer Academic Publishers, Dordrecht.)
- CROMER, D. T. & WABER, J. T. (1974). International Tables for X-ray Crystallography, Vol. IV, Table 2.2B. Birmingham: Kynoch Press. (Present distributor Kluwer Academic Publishers, Dordrecht.)
- DONOHUE, J. & GOODMAN, S. H. (1967). Acta Cryst. 22, 352-354.
- FRENZ, B. A. & OKAYA, Y. (1980). Enraf-Nonius Structure Determination Package. Enraf-Nonius, Delft, The Netherlands.
- JOHNSON, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
- McLaughlin, M. L., CRONAN, J. M. JR & FRONCZEK, F. R. (1988). Acta Cryst. C44, 1747–1749.
- Nordman, C. E. & Schmitkons, D. L. (1965). Acta Cryst. 18, 764-767.
- NORMAN, N. & POST, B. (1961). Acta Cryst. 14, 503-507.
- OLAH, G. A., SURYA, P. G. K. & LIANG, G. (1977). J. Org. Chem. 42, 661–666.
- SWEN-WALSTRA, S. C. & VISSER, G. J. (1971). J. Chem. Soc. Chem. Commun. pp. 82-83.

#### Acta Cryst. (1989). C45, 1100–1102

## Structure of 6-Tosyl-1,6,11,17,18-pentaazatricyclo[12.2.1<sup>1,4</sup>.1<sup>8,11</sup>]octadeca-2,4(17),8(18),9-tetraene

#### By F. H. CANO AND C. FOCES-FOCES

UEI de Cristalografia, Instituto Rocasolano, CSIC, Serrano 119, 28006 Madrid, Spain

(Received 12 October 1988; accepted 31 January 1989)

**Abstract.**  $C_{20}H_{25}N_5O_2S$ ,  $M_r = 399 \cdot 51$ , triclinic,  $P\overline{1}$ , a = 10.7373 (5), b = 9.4581 (5), c = 10.8159 (4) Å,  $\alpha = 74.012$  (4),  $\beta = 98.504$  (3),  $\gamma = 77.587$  (3)°, U = 1005.8 (1) Å<sup>3</sup>, Z = 2,  $D_x = 1.319$  Mg m<sup>-3</sup>, graphite-

0108-2701/89/071100-03\$03.00

monochromated Cu Ka radiation,  $\lambda = 1.5418$  Å,  $\mu = 1.599$  mm<sup>-1</sup>, F(000) = 424, T = 293 K, R = 0.059 for 2903 observed reflexions  $[I > 3\sigma(I)]$ . The pyrazole rings, which are quite similar, are inclined at different

© 1989 International Union of Crystallography

<sup>\*</sup>Lists of H-atom coordinates and isotropic thermal parameters, anisotropic thermal parameters, bond distances and angles involving H, structure-factor amplitudes, torsion angles and least-squares planes have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 51722 (21 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Sqaure, Chester CH1 2HU, England.